



Review of Edible Plants in Dumpsites: Risks of Heavy Metals Toxicity and Implications for Public Health

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Authors' contributions

This work was carried out in collaboration between all authors. Authors NAO, SEO, GOA and SOE wrote the first draft of the manuscript. Authors NAO, SEO and GOA managed the literature searches and author SOE edited the manuscript. All authors read and approved the final manuscript.

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Review Article

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ABSTRACT

Studies of dumpsites have revealed that the surrounding soils and water are contaminated with high threshold of heavy metals through anthropogenic inputs. In this review, the uptake and toxicity risks of these heavy metals by habitual edible plants at levels above threshold limit and the implications for public health have been discussed. Edible plants are plants with nutritional and medicinal potentials which can salvage numerous human and animal needs when taken. Edible plants like most other underutilized plants in dumpsites have developed mechanisms which enable them to not

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only survive but accumulate high level of toxic heavy metals due to high level of environmental metal load in the dumpsites. This ultimately could lead to high human and animal exposure to these toxic elements through food-chain/food-web or direct ingestion of soils. The toxic effects caused by excess concentrations of these heavy metals in living organisms vary considerably and present numerous clinical situations ranging from neurological disorder, cellular damage among others and death in extreme cases. This review suggest the urgent need for policy makers to regulate the use of dumpsites for arable farming and the dependence on edible plants in dumpsites to avert heavy metal poisoning in populations.

Keywords: Dumpsite soils; edible plants; heavy metals; toxicity; public health.

1. INTRODUCTION

The use of dumpsites for arable farming and the use of dumpsite compost soils for increasing the fertility of soils for the cultivation of vegetables, fruits and food crops in Nigeria and most developing countries of the world are serious issues of concern to the general public who want to know the safety of such food crops for human and animal consumption [1,2]. The sources of heavy metal to plants are their growth media (soil, air and nutrients) from where they are taken up by roots or foliage [3-5]. Some heavy metals such as Cu, Zn, Mn, and Fe have been reported to be essential to plants' growth and development. Others such as Cd, Hg, As, Cr etc have been shown not to play any functional role in plants at all concentrations [6,7]. The behaviour of heavy metals in soil-plant system is crucial to the understanding of the mobility, bioavailability and toxicity of the heavy metals in an ecosystem [3,8]. Plants growing on dumpsites can accumulate toxic metals at high concentration causing serious risk to human health when consumed [9-13].

The uptake of heavy metals by plants from soil varies due the variation of the soil physicochemical properties (pH, organic matter content, cationic exchange capacity, etc) and the plant species involved [4,5,8,9,14,15]. High load of heavy metals have been reported in crops grown in abandoned heavy metal polluted areas and dumpsites [12-14].

Heavy metals exceeding threshold limits and leading to environmental pollution are introduced into the environment from anthropogenic sources such as domestic and industrial wastes, and agricultural activities among others. Since these heavy metals are non-biodegradable, they tend to accumulate in the ecosystem and are absorbed by flora and fauna where they bio-magnify and cause varying degree of toxicities [15-18]. In Nigeria as in most African countries,

population explosion has led to the generation of large quantities of waste which are usually dumped in arable farmlands without recourse to the concomitant risks they may impose in soil and surrounding surface and ground water [19,20].

Food crops and vegetables are known to be the largest sources of nutrients for the developing world due to low income and poor standard of living as a result of population explosion [1,8]. The dependence of these populations on these cheaper sources of nutrients demand greater attention since reports have shown that the habitual environment such as dumpsites where these food crops and vegetables are grown influences their overall content as they may contain toxicants such as potentially harmful elements/metals [2,5-7]. Poor legislation and the use of such dumpsites for the cultivation of edible vegetables, fruits and food crops for human and animal consumption without routine assessment of the toxicological health risk of such practice is misleading.

This paper, therefore aims to review the studies of the plant-uptake of heavy metals in dumpsites and surrounding environment and point out the risk and health implications posed to humans and animals on consumption of plants cultivated on such soils.

2. BIOAVAILABILITY AND MOBILITY OF HEAVY METALS TO PLANTS

The health risks of refuse dumpsites have been assessed in various ways using total and bio-available forms of heavy metal content of the organisms living in or within the vicinity of the dumpsites [21,22]. Although total heavy metals may not completely give the predictive insight of the health risks, the chemical species of the metals as influenced by the physicochemical properties of the waste soils critically measure the overall risk of a refuse dumpsite [4-6,21,22].

Reports have that soil-plant transfer of heavy metals constitute serious risks since organisms depend on energy recycling via food chain/web for the sustainability of the ecosystem [21-26]. We have previously reported that there is a strong correlation between the percentage mobile phase of heavy metals in dumpsites and the soil physicochemical properties [3-5]. Dumpsites soils have been reported to contain anthropogenic input of heavy metals at levels above threshold limits when compared to soils from uncontaminated area and the bioavailability of these heavy metals in most dumpsites in Nigeria is well documented [3-5,9-13,24,27]. Our earlier research experiences in most of the dumpsites in South-East, Nigeria have shown that the available metal in the sequentially extracted fractions followed the general order: residual > reducible > exchangeable > acid soluble > oxidizable for Cd, residual > oxidizable > reducible > acid soluble > exchangeable for Cu, reducible > oxidizable > acid soluble > residual > exchangeable for Mn, residual > exchangeable > reducible > oxidizable > acid

soluble for Pb, residual > reducible > oxidizable > exchangeable > acid soluble for Zn, residual > exchangeable > reducible > oxidizable > acid soluble for Fe, residual > oxidizable > acid soluble > reducible > exchangeable for Ni and residual > oxidizable > acid soluble > reducible > exchangeable for Cr [3-5,8-13,27]. These reports were in line with the reports of other researchers carrying out similar studies [23,24,28-33]. Consistently, reports have shown that sequential extraction of heavy metals in dumpsites soils indicated higher percentages (%) of the non-residual fraction for most metals except Cu. Our study on numerous dumpsites soils in South-East, Nigeria (as shown Fig. 1) implicated higher percentage mobility for metals studied except Cu. These our earlier reports have shown that the overall order of the mobility and bioavailability of selected metals in most of the dumpsites studied follow this trend: Cd > Fe > Pb > Mn > Zn > Cr > Ni > Cu [3-5,8-13,34]. Similar findings on dumpsites and heavy metal polluted sites have been reported and well documented [23-26].

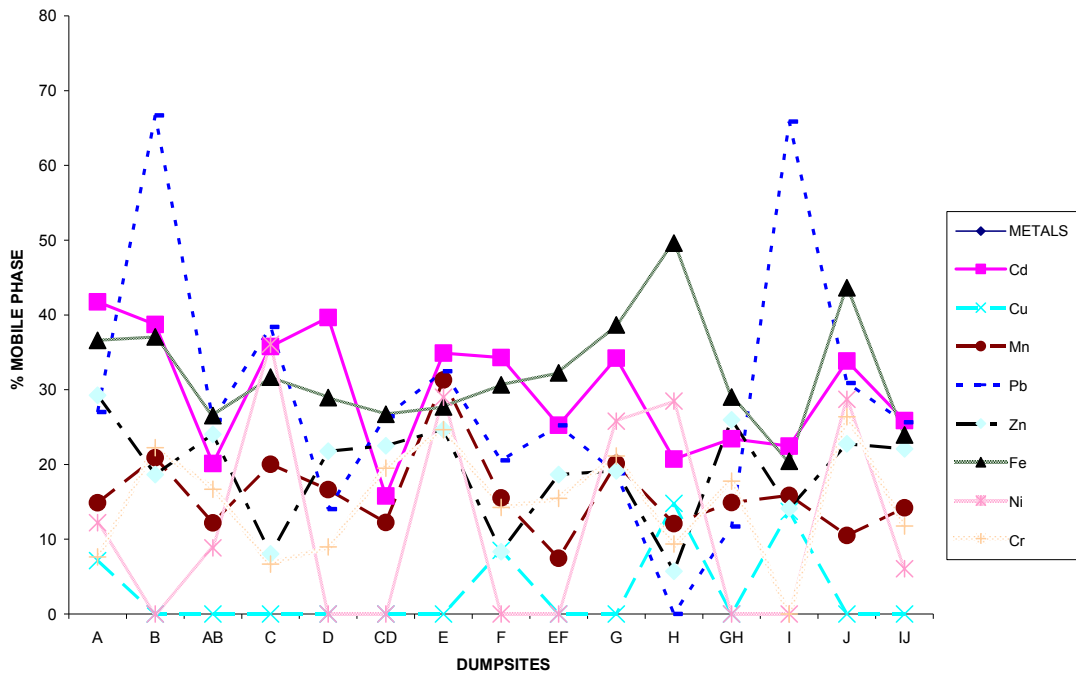


Fig. 1. Percentage (%) mobile phase of total extractable metals

Where A = Burrow pit-Aba, B = Ntiga-Isiala Ngwa, AB = Control site for Aba, C = Ubakala-Umuahia, D = Abia Tower-Umuahia, CD = Control site for Umuahia, E = Umuka-Okigwe, F = Ubahu-Okigwe, EF = Control site for Okigwe, G = Ishiagu-Enugu Junction, Ishiagu, H = Ntave-Ishiagu, GH = Control site for Ishiagu, I = Four corner-Enugu, J = Amaéchi-Enugu, IJ = Control site for Enugu.

Source: Obasi, 2012; Obasi et al., 2012a;b; Obasi et al., 2013a;b

3. HEAVY METAL UPTAKE BY PLANTS

Plants (roots, stem and leaves) grown in dumpsites and metal polluted soils have been reported to accumulate heavy metals far beyond the recommended threshold limits when compared to uncontaminated soils. The accumulation of these metals has been shown to vary greatly among plants species and uptake of these metals to the various plant parts has been demonstrated to be primarily dependent on the plant species, its inherent controls and the soil quality [3-5,8-14,35-42].

Soil-plant transfer indices of plants species such as translocation factors (TF), biological concentration factors (BCF) and biological accumulation coefficient (BAC) values greater than unity (>1) had been used to evaluate their potentials for phytoextraction, and phytostabilization [43-45]. Studies in Nigeria have shown that some edible and medicinal plants such as *Amaranthus hybridus*, *Talinum triangulare*, *Carica papaya*, *Ipomea batatas*, *Luffa aegyptica*, *Telferia occidentalis* among others that grows in dumpsites have Translocation Factor (TF), Biological Concentration Factor (BCF) or Biological Accumulation Coefficient (BAC) values greater than one (> 1) for metals such as Cd, Cu, Mn,

Pb, Zn, Fe, Ni and Cr among others [3-5,8-13, 27,34]. For example, our findings from dumpsites in Ibbi community, Afikpo North, Ebonyi State, South-East, Nigeria (Figs. 2-4) showed phytoremediation quotient greater than unity (> 1) for some metals. This suggests that these plant species have developed tolerance mechanisms to adapt to the heavy metals stress in their habitual environment and as such have potentials to accumulate these metals in their edible harvestable parts as documented by numerous researchers [8-13,35-42]. The implications are that these metals which have been accumulated by plants might be transferred through food chain and food web within the ecosystem to human and animals where they might pose serious threat to health.

4. HEAVY METALS AND PUBLIC HEALTH

At low and optimal concentrations, heavy metals are essential to the maintenance of various biochemical and physiological functions in living organisms. However, they exceed certain threshold limits, the constitute nuisance to flora and fauna. The toxicity of heavy metals arises from the fact that some of their chemical properties make them to avert control mechanisms such as homeostasis, transport, compartmentalization and binding of required cell

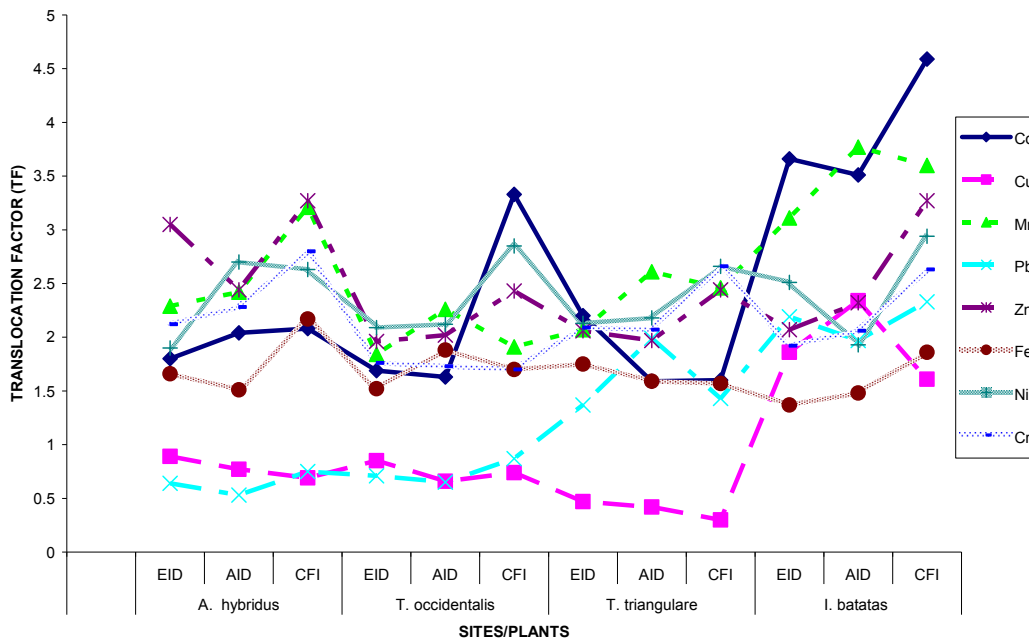


Fig. 2. Translocation factor (TF) of plants for all the metals in the studied sites

EID= Eluuogo Ibbi dumpsite, AID = Agboogo Ibbi dumpsite, CFI = Control Farmland Ibbi

Source: Obasi et al., 2015a

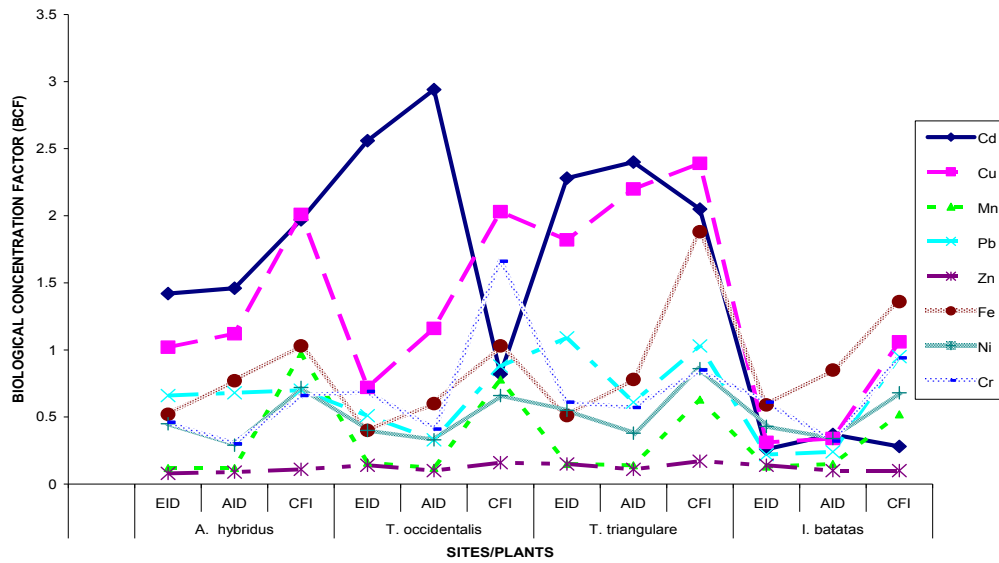


Fig. 3. Biological concentration factor (BCF) of plants for all the metals in the studied sites
 EID= Eluuogo Ibij dumpsite, AID = Agboogo Ibij dumpsite, CFI = Control Farmland Ibij
 Source: Obasi et al., 2015a

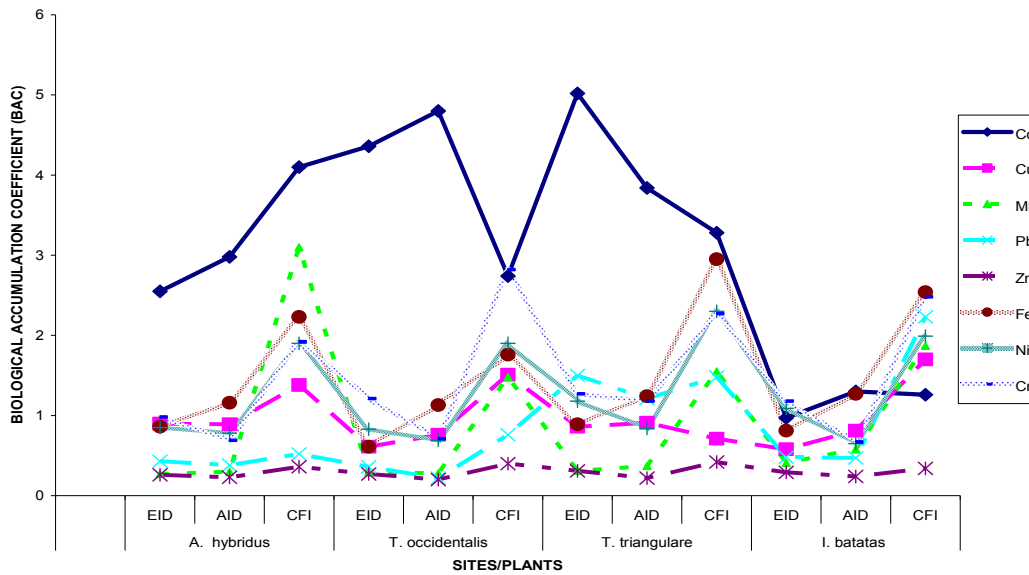


Fig. 4. Biological accumulation coefficient (BAC) of plants for all the metals in the studied sites
 EID= Eluuogo Ibij dumpsite, AID = Agboogo Ibij dumpsite, CFI = Control Farmland Ibij
 Source: Obasi et al., 2015a

constituents devised by living organisms for their protections. Thus, these chemical properties enable heavy metals to form coordinated molecular binding with protein and numerous biological molecules and as such replace or substitute the correct molecule from their natural binding sites leading to several toxicities causing malfunctioning of cells. The effects of toxicity of

heavy metals span across varying degrees of pathological conditions ranging from mutation to DNA damage at the molecular level to carcinogenicity, teratogenicity and metabolic disorders among others [46]. Several factors such as absorbed dose, route of exposure, duration of exposure (acute or chronic) and oxidative states determine to a large extent the

toxic effects of different heavy metals. The toxicities of heavy metals cover a wide facet because they are non-biodegradable and tend to recycle along the ecosystem [47,48]. Reports have shown that edible plants parts contain arsenic, cadmium, chromium, copper, lead, nickel, iron, manganese and zinc among others in varying amounts depending on the environment where the plants grew. These metals when absorbed from their habitual soil environment are transferred to other organisms in the ecosystem where they biomagnify and cause risks for human health and the environment [49-51].

The toxicity of heavy metals to humans and animals arise from various mechanisms involving binding to important proteins causing cellular damage and generation of reactive species.

Heavy metals have been reported to form intermediate compounds of high toxicities [49]. For example, inorganic arsenic compounds have been shown to be methylated by bacteria, algae, fungi and humans to give intermediate products (monomethylarsonic acid (MMA) and dimethylarsinic acid (DMA)) which highly toxic product that cause arsenic-induced carcinogenesis [52]. Similarly, lead metal toxicity have been shown to be due to its ability to replace other bivalent cations like Ca^{2+} , Mg^{2+} , Fe^{2+} and monovalent cations like Na^+ in various biological and metabolic processes. This ionic mechanisms causing lead toxicity manifest in various biological processes such as cell adhesion, intra- and inter-cellular signaling, protein folding, maturation, apoptosis, ionic transportation, enzyme regulation, and release of neurotransmitters [49]. For example Flora et al. [46] have demonstrated that lead can substitute calcium even in picomolar concentration and as such affect protein kinase C, which regulates neural excitation and memory storage.

Mercury toxicity is wide spread and has been reported to be potential neurotoxins [53]. Cadmium on the other hand has been shown to be hepatotoxic and nephrotoxic via its association with cysteine-rich protein such as metallothionein. Also, reports have indicated that when cadmium bind with cysteine, glutamate, histidine or aspartate ligands, they cause iron deficiency [54]. The replacement of zinc by cadmium due to same oxidation states in metallothionein, inhibit it from acting as a free radical scavenger within the cell [49,53].

The ionic specie of Cr(vi) and biological reductants such as thiols and ascorbate have been shown to produce reactive oxygen species (superoxide ion, hydrogen peroxide, and hydroxyl radical). This highly reactive ionic Cr specie like most other heavy metals ionic species cause oxidative stress in the cell and as such cause damage to DNA and proteins [55]. In the same way, free iron has been reported to cause lipid peroxidation, which may cause severe damage in mitochondria, microsomes and other cellular organelles [56]. The attack of heavy metals on a cell and the balance between reactive oxygen species (ROS) production and their subsequent defense by antioxidants is shown in Fig. 5 [49]. Iron toxicity on cells manifests as iron mediated tissue damage which arises from cellular oxidization and reduction mechanisms that affects intracellular organelles such as mitochondria and lysosomes. Also, excess intake of iron has been associated with the release of a wide range of free radicals. These free radicals have been shown to attack DNA, leading to cellular damage such as mutation and malignant transformations which in turn cause an diverse diseases [57].

The toxic effects of heavy metal cannot be over emphasized. The effects range from lowering energy levels to causing damage to the functionality of vital organs such as the brain, lungs, kidney, liver, blood composition among others. Long-term exposure to heavy metals and their compounds have been reported to cause gradual and progressive physical, muscular, and neurological degenerative processes. These processes initiate diseases such as multiple sclerosis, Parkinson's disease, Alzheimer's disease and muscular dystrophy and ultimately cancer on a repeated long-term exposure [58]. Concentration and duration induced effects of heavy metals varies. For example, lower and shorter levels of arsenic exposure has been shown to manifest clinically in nausea and vomiting, reduced production of erythrocytes and leukocytes, abnormal heart beat, etc. while long-term exposure have been implicated in the formation of skin lesions, internal cancers, neurological problems, pulmonary disease, peripheral vascular disease, hypertension and cardiovascular disease and diabetes mellitus [59]. Mazumder [60] reported that chronic arsenicosis results in many irreversible changes in vital organs with no effective treatment and high mortality rate.

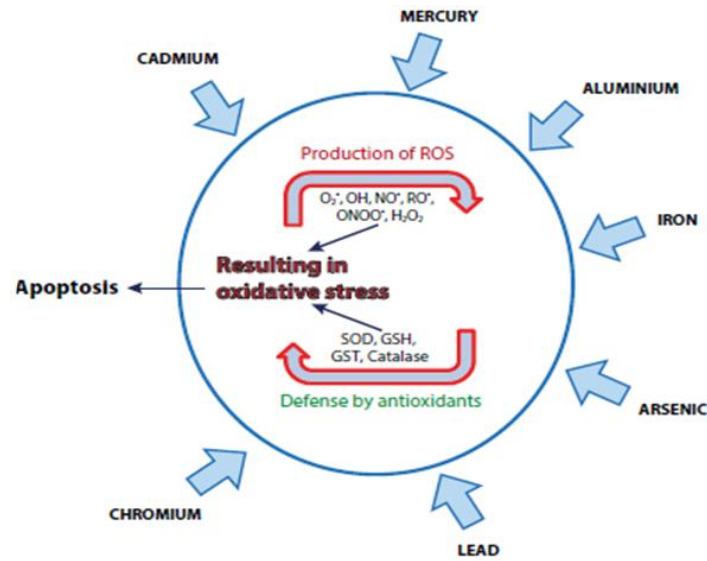


Fig. 5. The Effects of heavy metals on a cell and the balance between ROS production and the subsequent defense presented by antioxidants

Source: Jaishankar et al., 2014

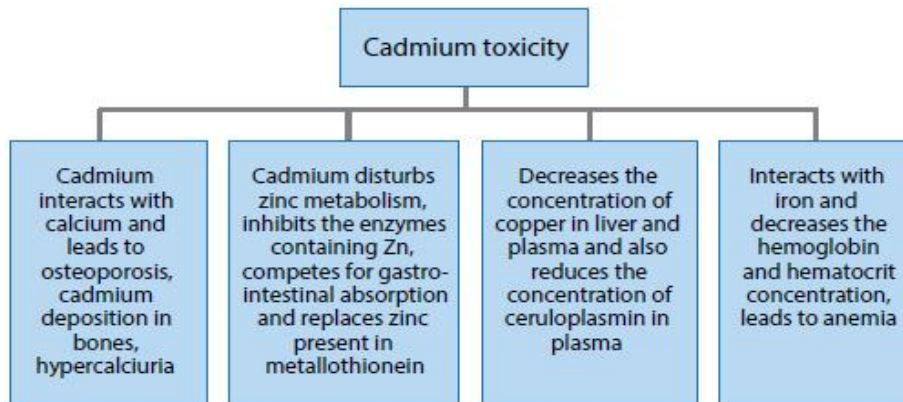


Fig. 6. Values of cadmium toxicity

Source: Flora et al., 2008

Exposure frequency and concentration of heavy metals affect their toxicities. For example, acute exposure heavy metals have been implicated in loss of appetite, headache, hypertension, abdominal pain, renal dysfunction, fatigue, sleeplessness, arthritis, hallucinations and vertigo. Chronic exposure of these metals on the other hand have been associated with mental retardation, birth defects, psychosis, autism, allergies, dyslexia, weight loss, hyperactivity, paralysis, muscular weakness, brain damage, kidney damage and may even cause death [61]. Reports have shown that exposure to higher amounts of chromium compounds in humans causes inhibition of erythrocyte glutathione

reductase, which interferes with methemoglobin in hemoglobin formation [62,63]. Other effects of chromate compounds include induction of DNA damage in different ways, for example: formation of DNA adducts, chromosomal aberrations, sister chromatid exchanges, alterations in replication and transcription of DNA leading to various forms of mutation with clinical manifestations in cancer and abnormal cells formation [64,65]. Excess iron uptake is a serious problem as it increases the risk of cancer while reports have shown that cadmium interacts with essential nutrients through which it causes its toxicity effects as summarized in Fig. 6 [46].

5. CONCLUSION

In this review, the effects of heavy metals accumulation by edible plants grown on dumpsites soils on public health were discussed. It was shown that plants grown on dumpsites accumulate heavy metals at levels beyond the threshold limit partly due to the high levels of heavy metals in the dumpsite soils and the ability of these plants to develop tolerance mechanisms in this habitual environment. These heavy metals can be transferred through feeding relationship (food chain and food web) to humans and animals where they exert numerous toxicities in humans and as such calls for public health concern. There is need for policy makers and environmentalist to put forward and enforce legislations guiding the management and disposal of domestic and industrial solid wastes and cultivation of food crops and vegetables in farmland within the vicinity of dumpsites. National and international co-operation is vital for framing appropriate tactics to prevent heavy metal toxicity arising from the use of dumpsites for arable farming and the use of dumpsites habitual plants for nutritional and medicinal purposes.

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COMPETING INTERESTS

Authors have declared that no competing interests exist.

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